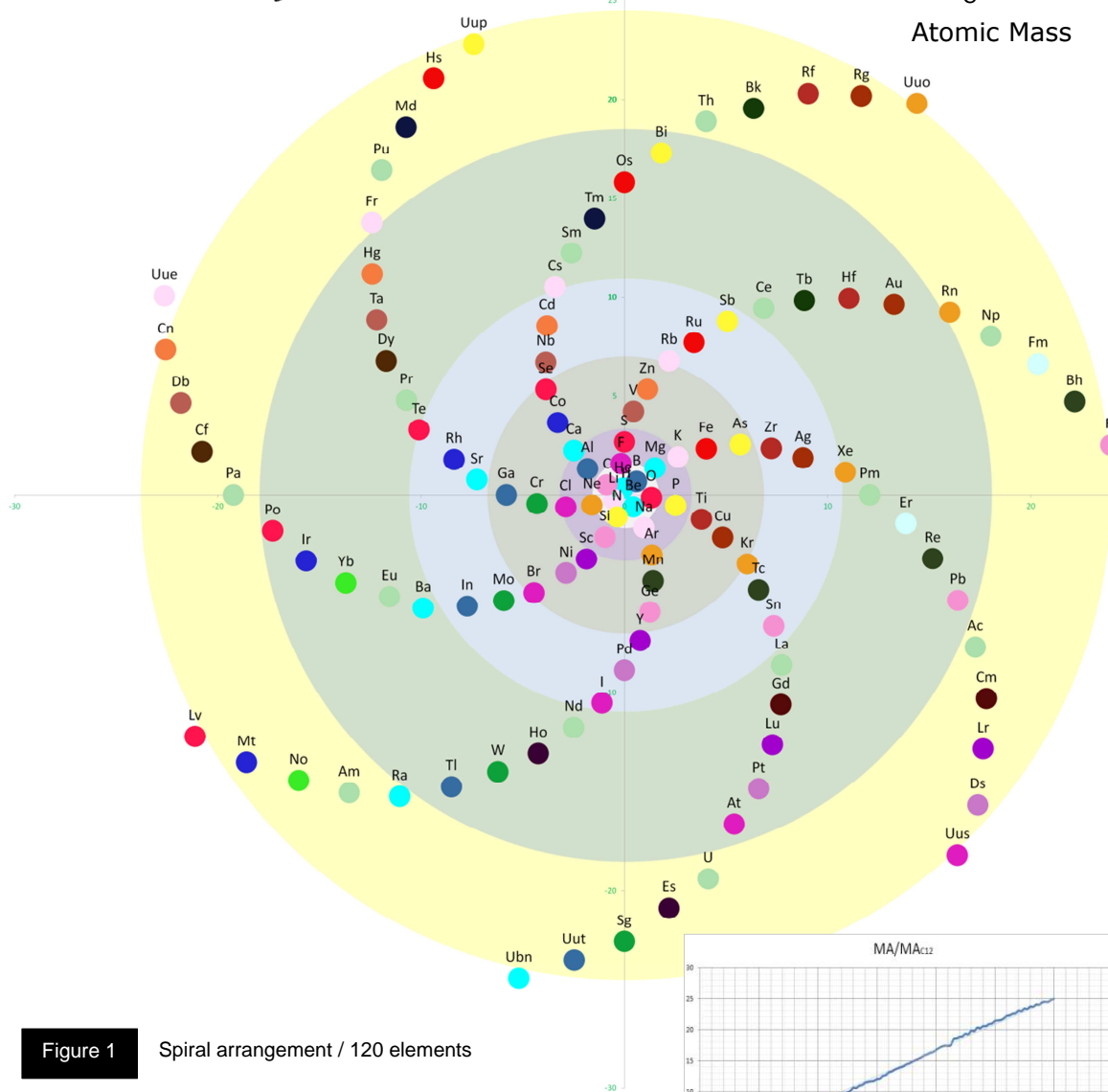


# Arrangement of Elements 7<sup>th</sup> order & Element Sequences, Periodic Table

– Olivier Joseph © –

7<sup>th</sup> order / 120 elements  
3<sup>rd</sup> Heptale angle  $\frac{\pi}{7}$

MA  
MA<sub>C<sup>12</sup></sub>  
Atomic Mass



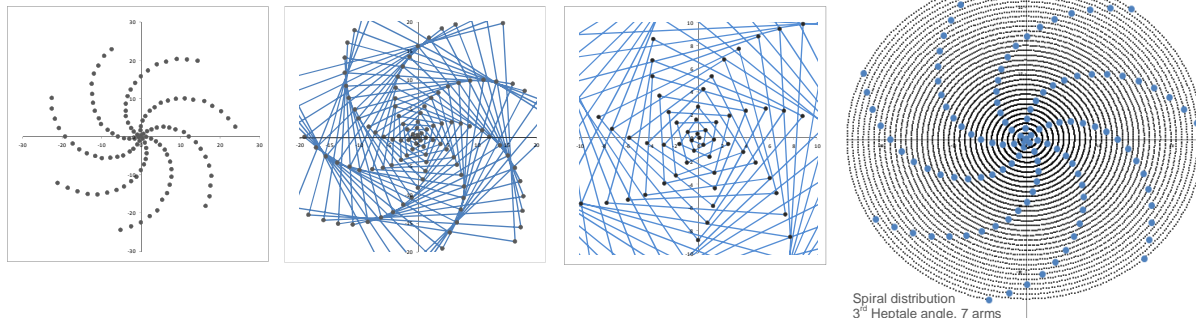
Atomic Number z

Element	z	beta	alpha	gamma	MA	10 <sup>6</sup> MA/MA <sub>C<sup>12</sup></sub>	MA/MA <sub>C<sup>12</sup></sub>
H	1	0	0.084	0	1.008	1.21319345	0.083930058
He	2	102	-0.009	0.326	4.003	2.15429701	0.33365579
Li	3	204	-0.528	-0.235	6.94	3.783134573	0.57785178
Be	4	306	0.441	-0.607	9.012	5.62826948	0.750374688
B	5	408	0.602	0.669	10.81	7.944805398	0.900083264
C	6	510	-0.866	0.5	12.01	10.57174173	1.204146645
N	7	612	-0.36	-1.109	14.01	14.67330326	1.166527893
O	8	714	1.325	-0.139	16	21.48934347	1.33223147
F	9	816	-0.165	1.573	19	38.1957452	1.582014988
Ne	10	918	-1.598	-0.519	20.18	47.89238275	1.680266445
Na	11	1020	0.957	-1.658	22.99	62.08014876	1.914238135
Mg	12	1122	1.504	1.354	24.31	105.7174173	2.024146645
Al	13	1224	-1.817	1.32	26.98	176.3848509	2.246461282
Si	14	1326	-0.951	2.137	28.09	218.2148305	2.338884263
P	15	1428	2.428	-2.522	30.97	279.0394643	2.57868448
S	16	1530	-25.15	2.669	32.04	407.1374007	2.699421132
Cl	17	1632	-2.887	-0.614	35.45	494.7607212	2.951706911
Ar	18	1734	1.353	-3.039	39.95	620.2871	3.236394671
K	19	1836	2.634	1.914	39.1	1801.442135	3.255620316
Ca	20	1938	-2.48	2.233	40.08	2173.799696	3.337218984
Sc	21	2040	-1.872	-3.242	44.96	5540.475568	3.743547044
Ti	22	2142	3.791	-1.232	47.87	9679.326265	3.985845129
V	23	2244	0.443	4.218	50.94	17436.74617	4.241465445
Cr	24	2346	-4.306	-0.453	52	21366.09864	4.329275229
Mn	25	2448	1.414	-4.351	54.94	37542.33082	4.574521232
Fe	26	2550	4.027	2.325	55.85	44698.343	4.650291424
Co	27	2652	-3.283	3.646	58.93	80676.00416	4.90674438
Ni	28	2754	-2.872	-3.953	58.69	77047.94015	4.886761032
Cu	29	2856	4.834	-1.523	63.55	156524.7561	5.254238113
Zn	30	2958	1.132	5.235	65.38	277941.3215	5.443798826
Ga	31	3060	-5.805	-9.15	69.72	638502.151	5.805162365
Ge	32	3162	1.257	-9.915	72.63	1115476.562	6.04746043
As	33	3264	5.699	2.537	74.92	173053.707	6.238134888
Se	34	3366	-3.864	5.319	78.96	375423.082	6.574521232
Br	35	3468	-4.452	-4.944	79.9	4495617.387	6.652789342
Kr	36	3570	6.043	-3.489	83.8	949519.596	6.977518734
Rb	37	3672	2.199	6.768	85.47	1307888.99	7.116569252
Sr	38	3774	-7.256	-0.763	87.62	19759005.49	7.295587101
Y	39	3876	1.774	-7.362	88.91	25292834.49	7.402397502
Zr	40	3978	7.224	2.347	91.22	3938577.61	7.595332719
Nb	41	4080	-3.868	6.7	92.91	54456946.87	7.79053289
Mo	42	4182	-5.938	-5.346	95.96	97725995.7	7.990008326
Tc	43	4284	6.601	-7.786	98.91	14449564.2	8.198866778
Ru	44	4386	3.423	7.688	101.07	2630777.7	8.415487094
Rh	45	4488	-8.381	1.782	102.91	37041874.5	8.588692756
Pd	46	4590	-4.14	-8.861	106.42	72902104.9	8.860949208
Ag	47	4692	8.785	1.867	107.87	95869824.6	8.981681932
Cd	48	4794	-3.807	8.551	112.41	228928704.1	9.35970029
In	49	4896	-7.734	-5.619	114.82	363384487	9.56036361
Sn	50	4998	7.345	-6.614	118.71	7660605784	9.884263114
Sb	51	5100	5.069	8.78	121.76	1374732468	10.1821815
Te	52	5202	-10.1	3.283	127.6	42119150210	10.6244796
I	53	5304	-1.104	-10.51	126.9	36829416448	10.56619484
Xe	54	5406	10.87	1.143	131.29	85452262062	10.93172355
Cs	55	5508	-3.42	10.52	132.91	116577411	11.06611116
Ba	56	5610	-9.908	-5.717	137.33	272043617	11.43486378
La	57	5712	7.291	-12.62	138.91	316824611	11.56891948
Ce	58	5814	6.858	9.439	140.12	444456411	11.66694421
Pr	59	5916	-10.72	4.772	140.91	540409411	11.73272273
Nd	60	6018	-2.497	-11.75	144.24	102327412	12.0099167
Pm	61	6120	12.07	4E-14	145.91	118378412	12.07327227
Sm	62	6222	-2.603	12.25	150.36	330801412	12.51956703
Eu	63	6324	-11.56	-5.146	151.96	449562412	12.65278934
Gd	64	6426	7.696	-10.59	157.25	123953413	13.09325562
Tb	65	6528	8.855	9.834	158.93	171056413	13.23313905
Dy	66	6630	-11.72	6.765	162.5	33915413	13.5309134
Ho	67	6732	-4.244	-13.06	164.93	540409413	13.73272273
Er	68	6834	13.85	-1.456	167.26	844749413	13.92672773
Tm	69	6936	-1.47	13.99	168.93	116557414	14.06577852
Yb	70	7038	13.7	-4.453	173.05	256346414	14.40882598
Lu	71	7140	7.291	-12.62	174.97	37041874	14.56891948
Hf	72	7242	11.04	9.944	178.49	7274145414	14.86178185
Ta	73	7344	-12.19	8.856	180.95	116577415	15.06661116
W	74	7446	-6.226	-13.98	183.84	202882415	15.30724396
Re	75	7548	15.17	-3.224	186.21	31958415	15.90457952
Os	76	7650	5E-14	15.84	190.23	690718415	15.83990058
Ir	77	7752	-15.66	-3.328	192.22	101157416	16.00499584
Pt	78	7854	6.607	-14.84	195.08	175037416	16.24313072
Au	79	7956	13.27	9.64	196.97	251478416	16.40049958
Hg	80	8058	-12.41	11.18	200.59	53402416	16.70191507
Tl	81	8160	-8.509	-14.74	204.38	104108417	17.01748543
Pb	82	8262	16.41	5.331	207.2	178768417	17.25228976
Bi	83	8364	1.819	17.31	208.98	251478417	17.40049958
Po	84	8466	-17.21	-1.819	209	254444417	17.40216486
At	85	8568	5.403	-16.63	210	307944417	17.65528811
Rn	86	8670	16.01	9.242	222	30528418	18.48459612
Fr	87	8772	-12.42	13.8	223	369709418	18.56786012
Ra	88	8874	-11.06	-15.22	226	657131418	18.81765196
Ac	89	8976	17.27	-7.688	227	796005418	18.9009159
Th	90	9078	4.017	18.9	232.04	209202419	19.32056619
Pa	91	9180	-19.24	2E-13	231.04	172704419	19.23730225
U	92	9282	4.121	-19.39	238.03	659656419	19.1931724
Np	93	9384	18.03	8.026	237	541446419	19.7335537
Pu	94	9486	-11.94	16.44	244	207206420	20.316403
Am	95	9588	-13.54	-15.04	243	171056420	20.23313905
Cm	96	9690	17.81	-10.28	247	368294420	20.56619484
Bk	97	9792	6.355	19.56	247	368294420	20.56619484
Cf	98	9894	-20.78	1.85	251	732956420	20.89250821
Es	99	9996	2.193	-20.87	252	960538420	20.8251457
Fm	100	10098	20.35	6.613	257	250515421	21.3988343
Md	101	10200	-10.74	18.6	259	30458421	21.48209825
No	102	10302	-16.03	-14.43	259	367589421	21.5653622
Lr	103	10404	17.65	-12.82	262	65362421	21.81515404
Rf	104	10506	9.042	20.31	267	170402422	22.23147377
Sg	105	10608	-21.83	4.639	268	206413422	22.31473772
Dh	106	10710	-4E-13	-22.56	271	36888422	22.56452956
Bh	107	10812	22.15	4.709	272	44442422	22.64779351
Hs	108	10914	-9.381	21.07	277	115908423	23.06411324
Mt	109	11016	-18.59	-13.51	276	95688422	22.98048929
Uu	110	11118	17.39	-15.66	281	249557423	23.9716903
Nh	111	11220	11.66	20.18	280	206018423	23.31395268
Fl	112	11322	-21.67	3.333	285	537318423	23.70324891
Uu	113	11424	3.472	-23.32	284	443569423	23.64692891
Uu	114	11526	23.93	2.515	289	115898424	24.0628206
Uu	115	11628	-7.41	22.81	288	952029423	23.98016565
Lv	116	11730	-21.13	-12.2	293	249079424	24.39633639
Uu	117	11832	16.38	-18.19	294	301717424	24.47960033
Uuo	118	11934	14.39	19.8	294	301717424	24.47960033
Uue	119	12036	-22.63	10.08	297.3006	5.9676424	24.7758
Uub	120	12138	-1.94	-24.44	299.808	9.63829424	24.984

Figure 1 Spiral arrangement / 120 elements

$$H(n) = \frac{1}{12} \{ 2n^3 + 6n^2 + (7 - 3(-1)^n)n - 3(-1)^n - 9 \}$$

$n \geq 1, n \in \mathbb{N}$



Spiral distribution  
3<sup>rd</sup> Heptale angle, 7 arms

9<sup>th</sup> order / 220 elements  
 3<sup>rd</sup> Heptale angle  $\frac{2\pi}{7}$   
 [Extrapolation MA Lvl 1]

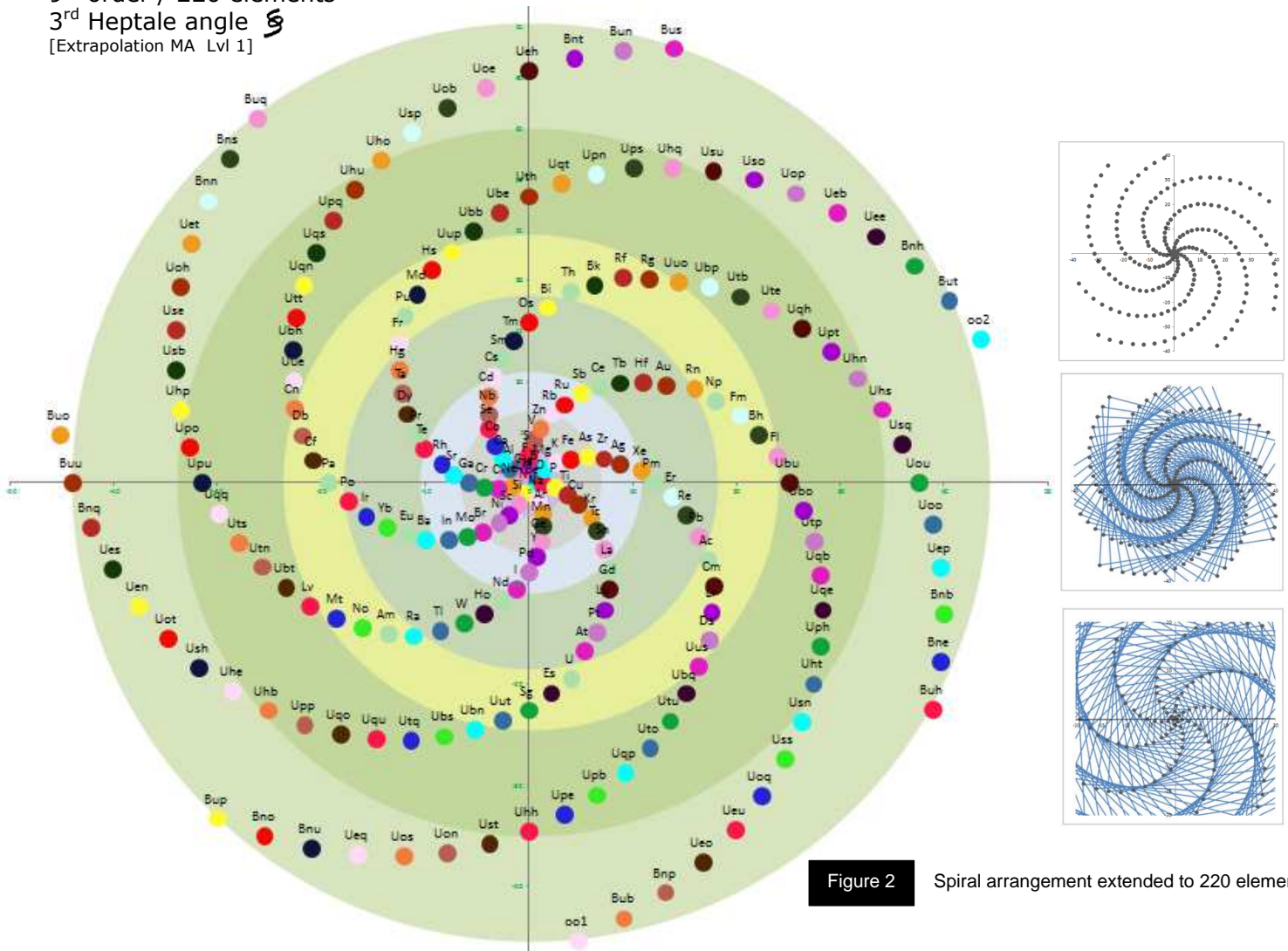
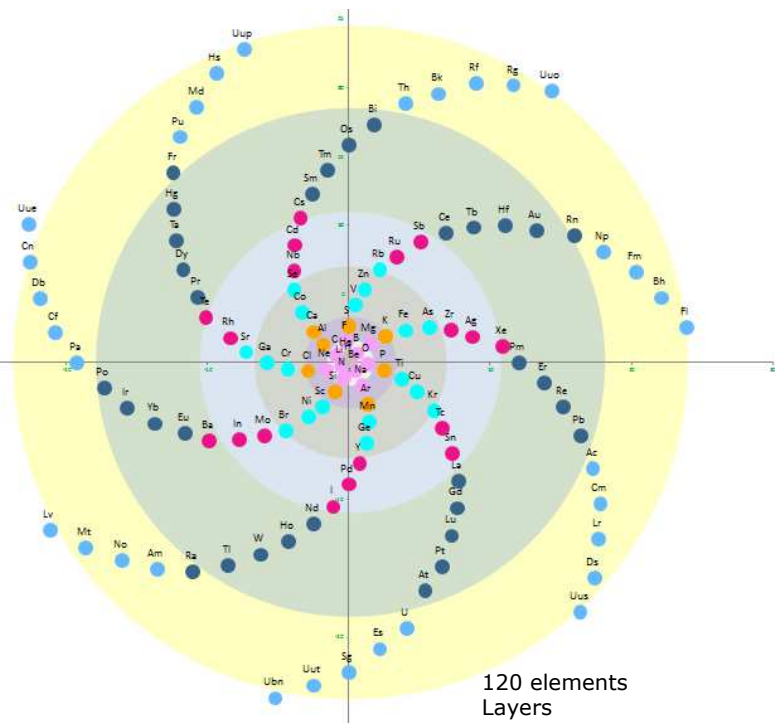
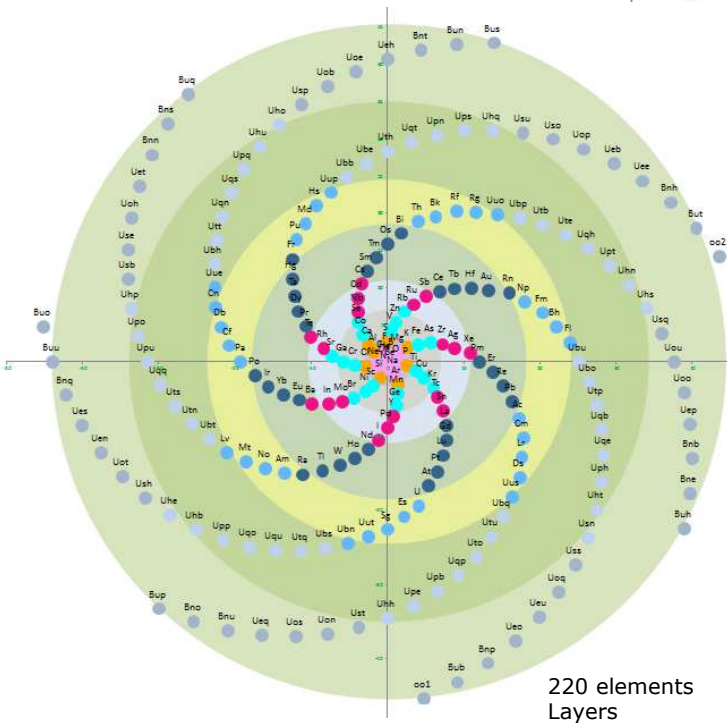
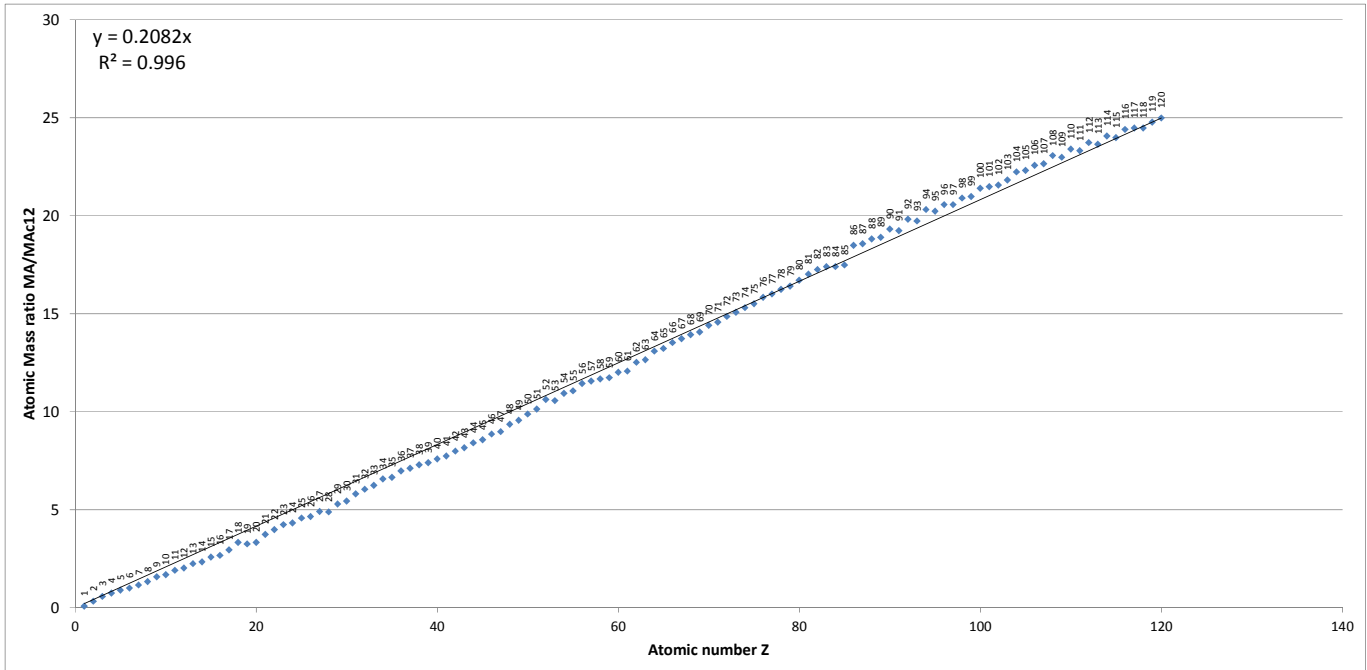


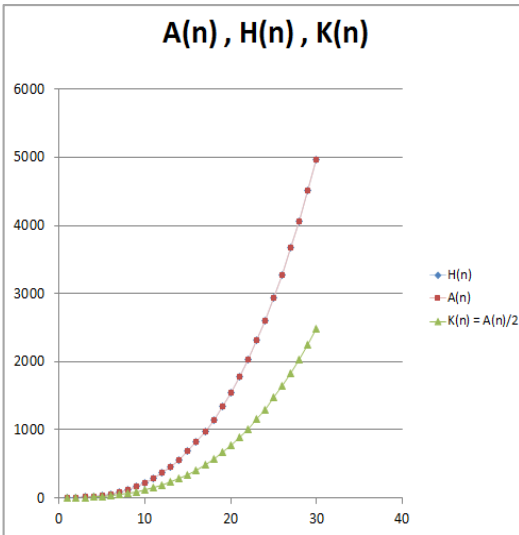
Figure 2 Spiral arrangement extended to 220 elements





**Figure 3** Atomic mass vs. Atomic number

Z = The Atomic Number, the Number of Protons in the Nucleus



n	H(n)	A(n)	K(n) = A(n)/2
1	1	0	0
2	3	2	1
3	11	10	5
4	19	18	9
5	37	36	18
6	55	54	27
7	87	86	43
8	119	118	59
9	169	168	84
10	219	218	109
11	291	290	145
12	363	362	181
13	461	460	230
14	559	558	279
15	687	686	343
16	815	814	407
17	977	976	488
18	1139	1138	569
19	1339	1338	669
20	1539	1538	769
21	1781	1780	890
22	2023	2022	1011
23	2311	2310	1155
24	2599	2598	1299
25	2937	2936	1468
26	3275	3274	1637
27	3667	3666	1833
28	4059	4058	2029
29	4509	4508	2254
30	4959	4958	2479

$$H(n) = A(n) + 1$$

$$H(n) = \frac{1}{12} \{ 2n^3 + 6n^2 + (7 - 3(-1)^n)n - 3(-1)^n - 9 \}$$

$n \geq 1$

$$H(2p) = \frac{1}{6} (8p^3 + 12p^2 + 4p - 6) \quad \text{Even numbers} \quad p \geq 1$$

$$H(2p + 1) = \frac{1}{6} (8p^3 + 24p^2 + 28p + 6) \quad \text{Odd numbers} \quad p \geq 0$$

$$A(n) = \frac{1}{12} \{ 2n^3 + 6n^2 + (7 - 3(-1)^n)n - 3(-1)^n - 21 \}$$

$n \geq 1$

$$A(2p) = \frac{2}{3} (2p^3 + 3p^2 + p - 3) \quad \text{Even numbers} \quad p \geq 1$$

$$A(2p + 1) = \frac{2}{3} p(2p^2 + 6p + 7) \quad \text{Odd numbers} \quad p \geq 0$$

$$K(n) = A(n)/2$$

$n \geq 1$

$$K(2p) = \frac{A(2p)}{2} \quad \text{Even numbers} \quad p \geq 1$$

$$K(2p + 1) = \frac{A(2p + 1)}{2} \quad \text{Odd numbers} \quad p \geq 0$$

$$K(2p + 1) = K(2p) + (p + 1)^2$$

$$H(2p + 1) = H(2p) + 2(p + 1)^2$$

**Table 1** H(n) sequence (Group IA) elements (H, Li, Na, K, Rb, Cs,...and beyond), and sub-sequences

1	2	3	4	5	6	7	8	...
1	3	11	19	37	55	87	119	...
H	Li	Na	K	Rb	Cs	Fr	Uue	...



Figures 1,2 and 4,5 show that the elements are notably arranged in a specific pattern (in a spiral form made up of 7 arms) when an angle approximately equal to  $\xi \approx 102$  degrees is set. It is to be noted that other angle values generate various interesting patterns with different but less remarkable distribution of elements.

With this specific angle of about 102 degrees, all the elements of a given group (corresponding to chemical elements with similar properties, such as Ne, Ar, Kr, Xe, Rn for the noble gases...) also become specifically distributed among each arm of the spiral in a notable arranged way, fairly close to a regular spiral pattern as well, curved in the opposite direction.

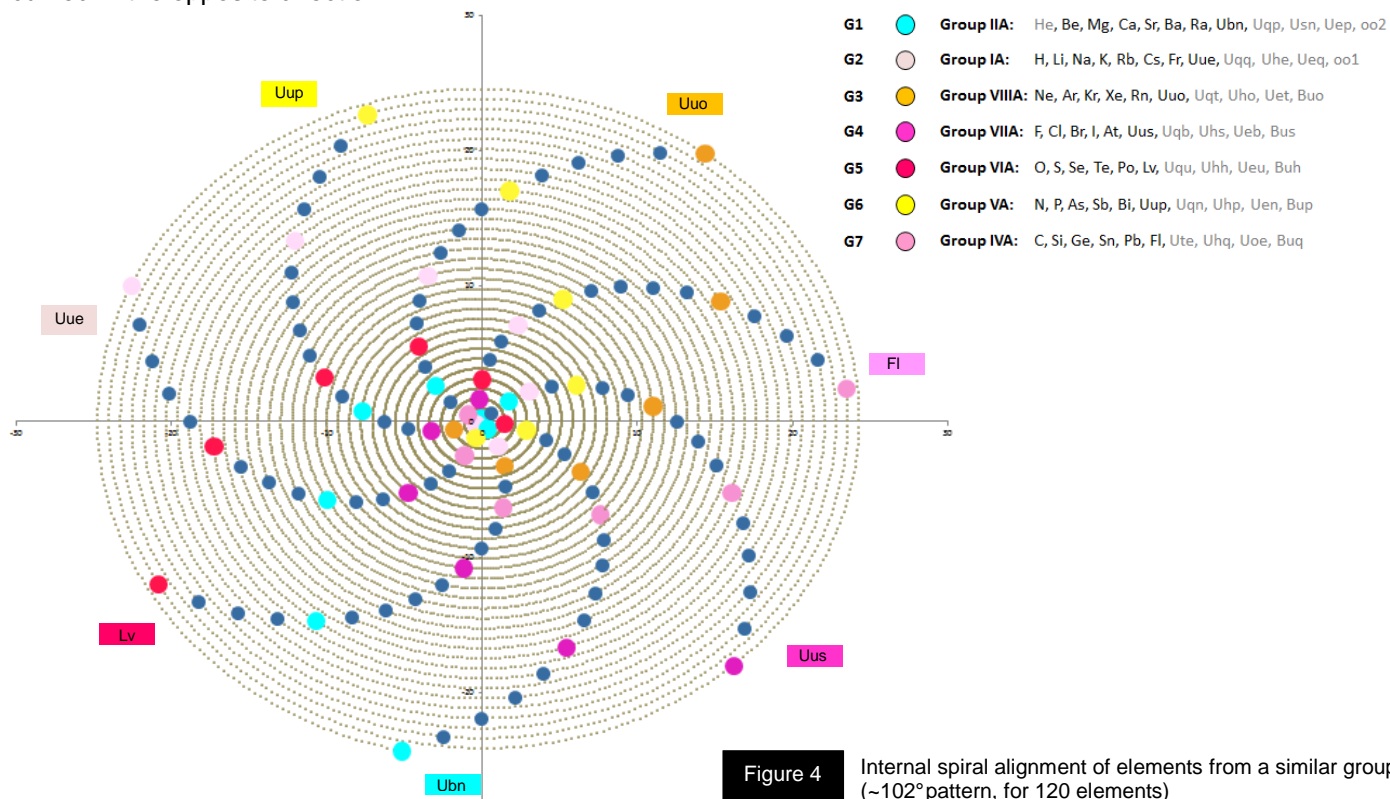


Figure 4 Internal spiral alignment of elements from a similar group (~102° pattern, for 120 elements)

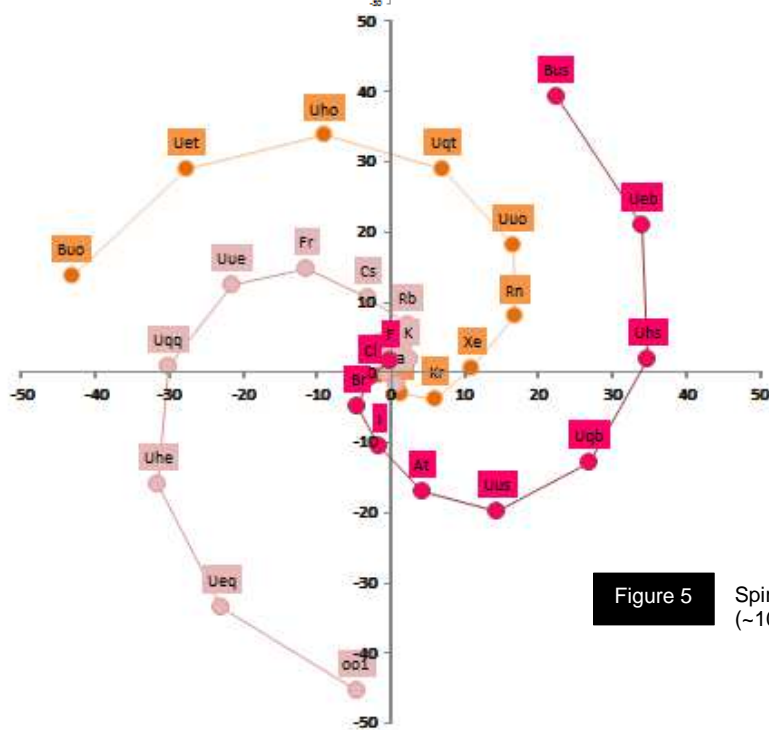


Figure 5 Spiral curve pattern per group (e.g. Group IA, VIIA, VIIIA) (~102° pattern, for extended 220 elements)

PHI Theor. Deviation 220 elements  
 3<sup>rd</sup> Heptale angle  $\xi$   
 Matching / Ratio:  $\varepsilon$   
 Slope:  $\log(\text{PHI})$

$$\text{PHI} = \frac{1 + \sqrt{5}}{2}$$

9<sup>th</sup> order / 220 elements  
 3<sup>rd</sup> Heptale angle  $\xi$   
 [Extrapolation MA Lvl 1]

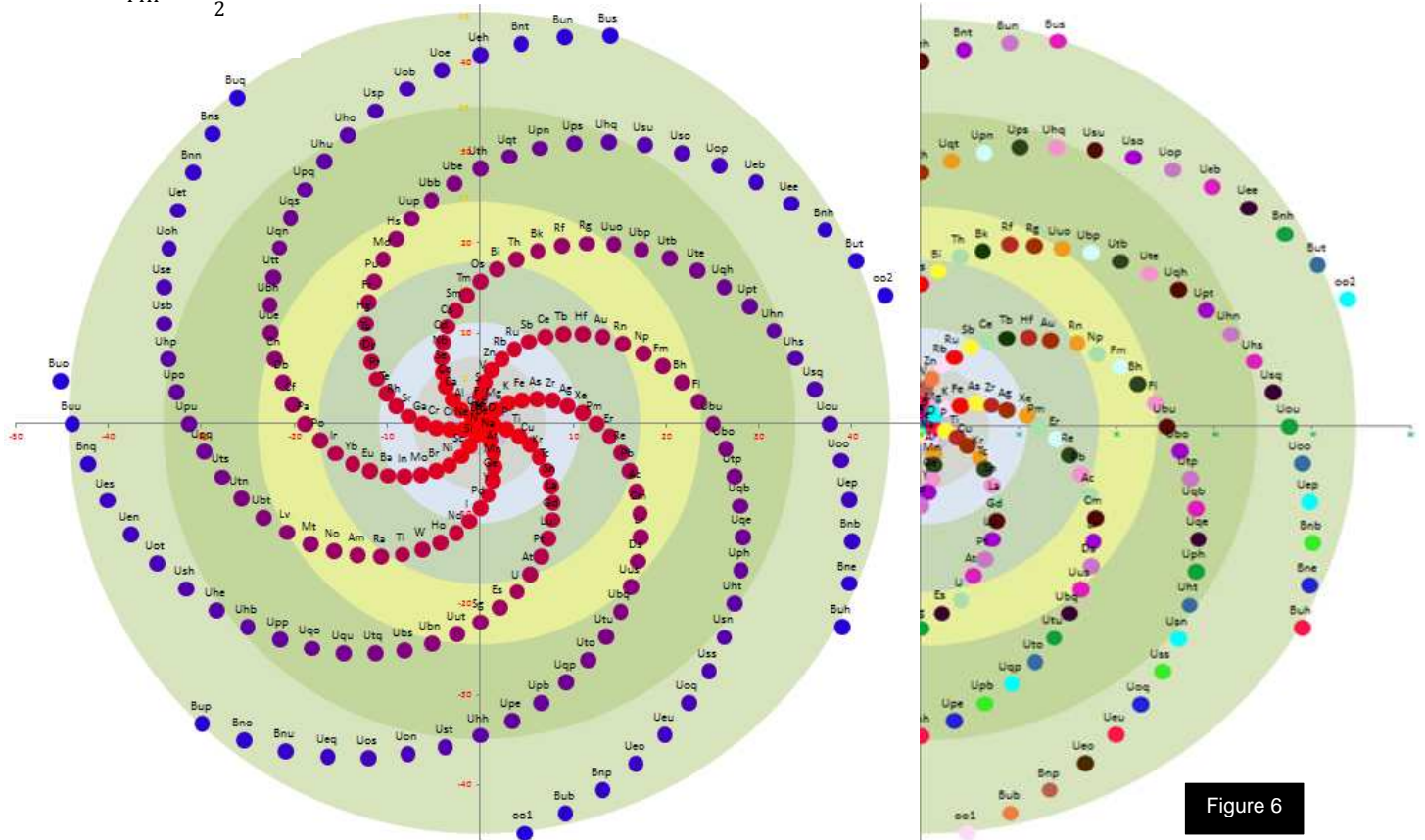
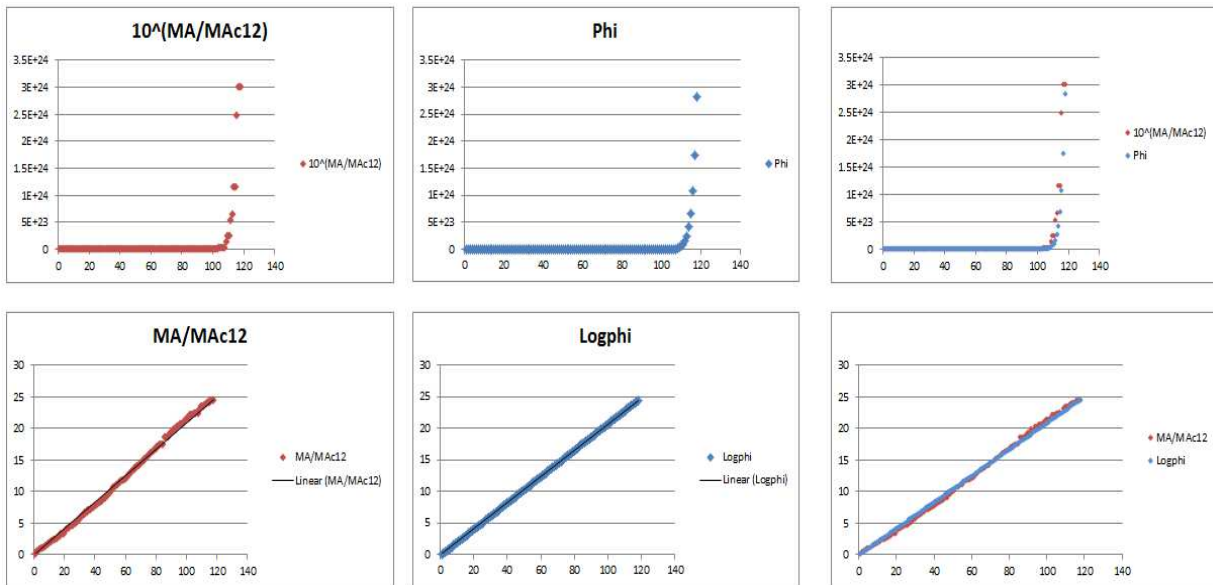


Figure 6

LogPHI vs. MA/MA<sub>c12</sub> - Matching comparison



$$Y = \log(\text{PHI}) \times \log(\text{PHI})$$

$$R^2 = 1$$

Figures 6 and 7 exhibit another feature of the arrangement with the very close match observed between the two arrangements, consisting of the MA/MA<sub>c12</sub> pattern as radial distance on one side and the LogPHI pattern

(PHI = golden mean  $\frac{1+\sqrt{5}}{2}$ ) as radial distance on the other side.

As depicted here below, both models appear to fairly overlap.

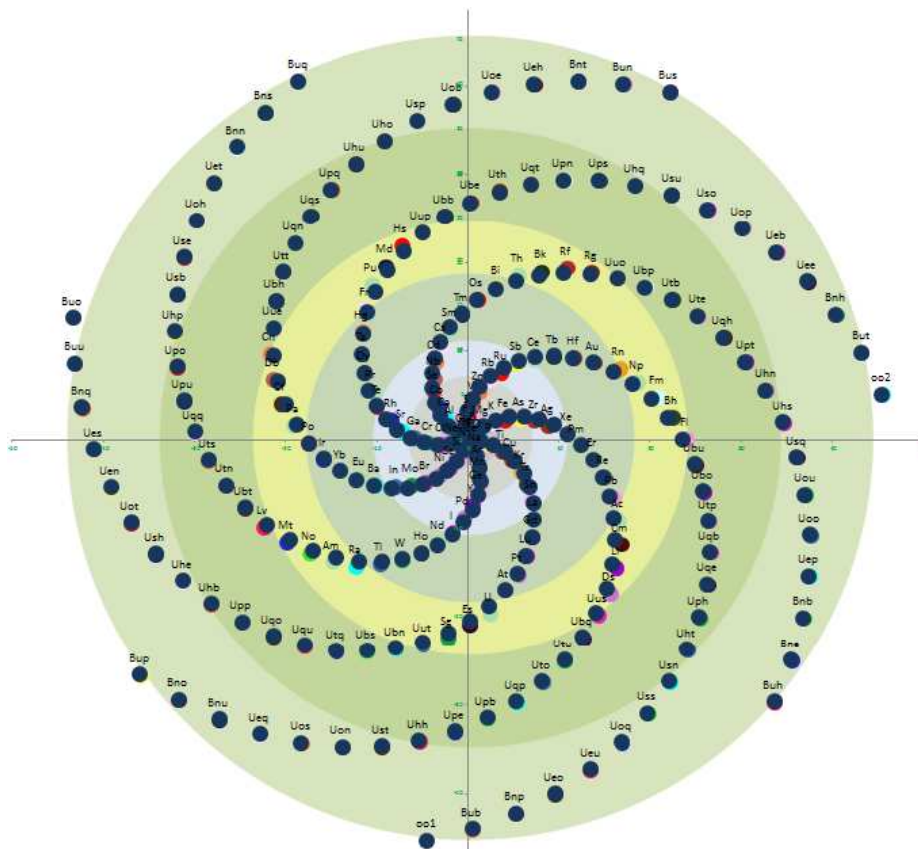


Figure 7 LogPHI (in upper position and blue color) vs. MA/MA<sub>c12</sub> overlapped patterns, showing very minor deviations.

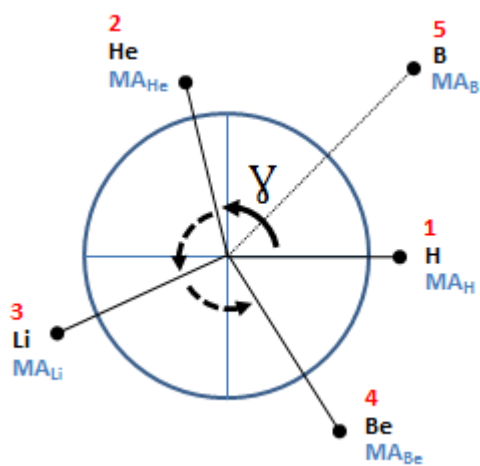
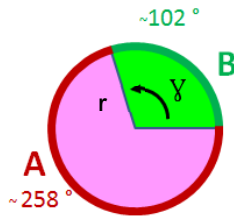


Figure 8 First five elements positioning, according to the constant angle  $\gamma = \frac{\pi}{5}$  spacing and MA/MA<sub>c12</sub> as the radial distance

## Best $\gamma$ angle and associated $\mathcal{K}$ ratio determination

Given  $\gamma$  angle, between two successive elements, characterized as follows, together with its associated  $\mathcal{K}$  ratio:



$$A + B = 2\pi r$$

$$B = \gamma r$$

$$\gamma_{Rad} = \gamma_{Deg} \cdot \frac{\pi}{180} = \frac{2\pi B}{A + B}$$

$$\mathcal{K} = \frac{A}{B} \quad (\text{called } \mathcal{K} \text{ ratio})$$

$$\gamma_{Rad} = \frac{2\pi}{\mathcal{K} + 1}$$

$$\gamma_{Deg} = \frac{360}{\mathcal{K} + 1}$$

leading to:

$$\mathcal{K} = \frac{2\pi - \gamma_{Rad}}{\gamma_{Rad}}$$

$$\mathcal{K} = \frac{360 - \gamma_{Deg}}{\gamma_{Deg}}$$

Among the pool of possible angle values, the best candidate angle  $\gamma_{Best}$  would presently be an angle that generates at least 7 spiral arms (close to a multiple of  $2\pi/7$  in theory) and providing the most possible optimized distribution of elements. Such a preferred angle is an irrational angle.

The angle of  $\pi$  rotations or a derived multiple of  $\pi$  (i.e.  $k * \pi * 360^\circ$ ) matches the requirements.

$$\gamma_{Best} = 2 * \pi * 360$$

This angle shall moreover be adequately expressed, by isolating its most valuable part with its remainder in a range of  $0^\circ$  to  $360^\circ$  (360 as the divisor). Therefore,  $\gamma_{Best}$  is equivalently evaluated to  $\gamma_{Deg}$ .

$$\gamma_{Best} = 2 * \pi * 360 = \gamma_{Deg} + 6 * 360 \quad (\gamma_{Deg} = \text{Modulo} [\gamma_{Best}, 360^\circ])$$

$$\gamma_{Deg} = 2(\pi - 3) * 360$$

$$\gamma_{Deg} \approx 101.9467106^\circ = \mathfrak{S} \approx 102^\circ$$

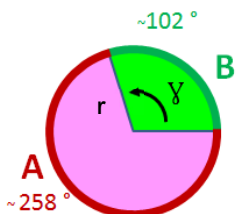
$$\gamma_{Rad} = \gamma_{Deg} \cdot \frac{\pi}{180}$$

$$\gamma_{Rad} = 4\pi(\pi - 3)$$

$$\gamma_{Rad} \approx 1.779305761$$

$$\mathcal{K} = \frac{7 - 2\pi}{2(\pi - 3)}$$

$$\mathcal{K} \approx -2.531256653$$



~102°..	28.3 %	2.531256653 = $\mathcal{K}_1 = \frac{7 - 2\pi}{2(\pi - 3)}$
~258°..	71.7 %	0.39506069 = $\mathcal{K}_2 = \frac{2(\pi - 3)}{7 - 2\pi}$



## Quadratic equation: $\mathcal{X}_1, \mathcal{X}_2$ solutions

Finding the second degree polynomial equation having its 2 roots as  $\mathcal{X}_1$  and  $\mathcal{X}_2$ , in the form of an equation and resulting from the equivalency between ratios.

B	A	$nA + mB$
1	$\frac{A}{B}$	$\frac{nA + mB}{A}$

$$\boxed{X^2 - nX - m = 0}$$



$$\boxed{\frac{A}{B} = \frac{nA + mB}{A}}$$

$$\mathcal{X} = \frac{A}{B}$$

Equation solved by:

$$\left\{ \begin{array}{l} X_1 = \frac{n + \sqrt{n^2 + 4m}}{2} = \frac{7 - 2\pi}{2(\pi - 3)} = \mathcal{X}_1 \\ X_2 = \frac{n - \sqrt{n^2 + 4m}}{2} = \frac{2(\pi - 3)}{2\pi - 7} = -\mathcal{X}_2 = -\frac{1}{\mathcal{X}_1} \end{array} \right.$$

$$\Rightarrow \left\{ \begin{array}{l} n = \frac{4\pi - 13}{2(\pi - 3)(2\pi - 7)} \\ m = 1 \end{array} \right.$$



$$\boxed{X^2 - \frac{4\pi - 13}{2(\pi - 3)(2\pi - 7)}X - 1 = 0}$$

$$X_1 = \mathcal{X}_1 = 2.531256653$$

$$X_2 = -\mathcal{X}_2 = -0.39506069$$

In other words, this means:



$$\boxed{X^2 = \Pi X + 1}$$

given

$$\boxed{\Pi = \frac{4\pi - 13}{2(\pi - 3)(2\pi - 7)}}$$

$$\approx 2.136195963$$

Or which can be rearranged to

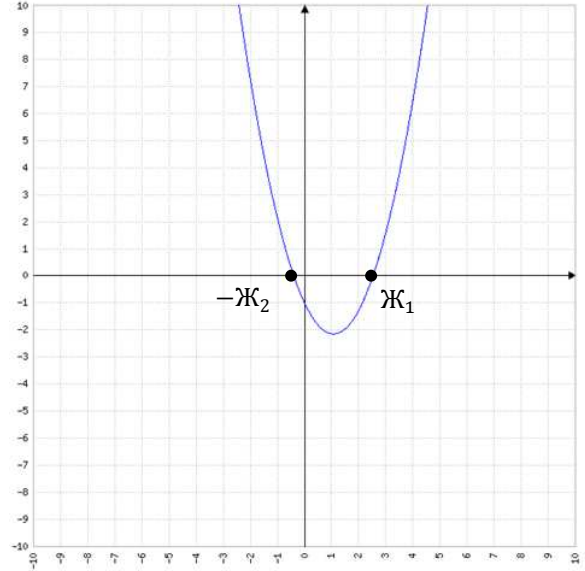
$$\boxed{\frac{1}{X} = X - \Pi}$$

Leading to  $\mathcal{X}$ :

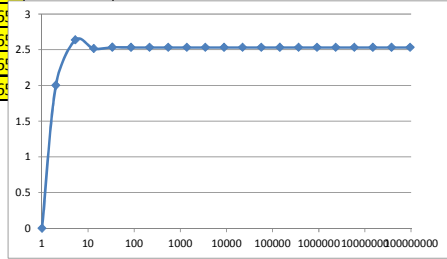


$$\boxed{\mathcal{X} = \Pi + \frac{1}{\Pi + \frac{1}{\Pi + \frac{1}{\Pi + \frac{1}{\Pi + \frac{1}{\Pi + \dots}}}}}}$$

0	u1= 1=B	1	0
1	u2= 2=A	2	2 u2/u1
2	u3= ΠA+B	5.272392	2.636195963 u3/u2
3	u4=Π(ΠA+B)+A	13.26286	2.515530434 u4/u3
4	...	33.60446	2.533726435 u5/u4
5	...	85.04858	2.530871563 u6/u5
6	...	215.2849	2.531316764 ...
7	...	544.9393	2.531247271 ...
8	...	1379.382	2.531258117
9	...	3491.57	2.531256424
10	...	8838.06	2.531256689
11	...	22371.4	2.531256647
12	...	56627.75	2.531256654
13	...	143339.4	2.531256653
14	...	362828.7	2.531256653
15	...	918412.6	2.531256653
16	...	2324738	2.531256653
17	...	5884509	2.531256653
18	...	14895201	2.531256653
19	...	37703577	2.531256653
20	...	95437431	2.531256653



Ratio properly converging to  $\mathcal{K}_1$  after a few iterations as n index increases



**$\mathcal{K}$  approximation, Sequences associated to  $\mathcal{K}$  :**

Specifying the recursive sequences:

$$X_n = \Pi + \frac{1}{X_{n+1}}$$

$n \geq 1$

$X_0 = \Pi$

Alternatively, the below recurrence relation can be expressed and approximates  $\mathcal{K}$  :

$$U_{n+2} = \Pi U_{n+1} + U_n$$

$n \geq 1$

$U_1 = 1$

$U_2 = \Pi$

....

$$\mathcal{K} = \lim_{n \rightarrow \infty} \left( \frac{U_{n+1}}{U_n} \right).$$

$\frac{U_{n+1}}{U_n}$  ratio properly converging to  $\mathcal{K}_1$  →

	Un	Un+1/Un
1 U1	1	
2 U2	2.136195963	2.60431781
3 U3	5.563333192	2.520173677
4 U4	14.02056587	2.532994013
5 U5	35.5140094	2.530985684
6 U6	89.88544937	2.531298948
7 U7	227.5269435	2.531250052
8 U8	575.9275875	2.531257683
9 U9	1457.821131	2.531256492
10 U10	3690.119202	2.531256678
11 U11	9340.638872	2.531256649
12 U12	23643.55425	2.531256654
13 U13	59847.90401	2.531256653
14 U14	151490.4052	2.531256653
15 U15	383461.096	2.531256653
16 U16	970638.4504	2.531256653
17 U17	2456935.035	2.531256653
18 U18	6219133.154	2.531256653
19 U19	15742222.17	2.531256653
20 U20	39847604.6	2.531256653
21 U21	100864514.3	2.531256653
22 U22	255313972.8	2.531256653
23 U23	646265192.1	2.531256653
24 U24	1635863067	2.531256653
25 U25	4140789272	2.531256653
26 U26	10481400394	2.531256653
27 U27	26531114479	2.531256653
28 U28	67157060035	2.531256653
29 U29	1.69992E+11	2.531256653
30 U30	4.30293E+11	2.531256653
31 U31	1.08918E+12	2.531256653
32 U32	2.757E+12	2.531256653

## U(k) sequence according to k index (using Binet forms):

As previously stated, the 2 roots  $X_1 = \mathcal{X}_1 = \mathcal{X}$  and  $X_2 = -1/\mathcal{X}$  satisfy the equation  $X^2 = \Pi X + 1$

$$\begin{aligned} X_1^2 &= \Pi X_1 + 1 & X_1^{k+2} &= \Pi X_1^{k+1} + X_1^k \\ X_2^2 &= \Pi X_2 + 1 & X_2^{k+2} &= \Pi X_2^{k+1} + X_2^k \end{aligned}$$

Thus, the sequence  $F_k = \alpha \mathcal{X}^k + \beta \left(-\frac{1}{\mathcal{X}}\right)^k$  satisfies the same recurrence relation

$$F_{k+2} = \Pi F_{k+1} + F_k$$

Indeed,

$$F_{k+2} = \alpha \mathcal{X}^{k+2} + \beta \left(-\frac{1}{\mathcal{X}}\right)^{k+2} = \alpha \Pi \mathcal{X}^{k+1} + \alpha \mathcal{X}^k + \beta \Pi \left(-\frac{1}{\mathcal{X}}\right)^{k+1} + \beta \left(-\frac{1}{\mathcal{X}}\right)^k = \Pi F_{k+1} + F_k$$

$$F_k = \alpha \mathcal{X}^k + \beta \left(-\frac{1}{\mathcal{X}}\right)^k$$

$$F_0 = \alpha + \beta$$

$$F_1 = \alpha \mathcal{X} - \beta \left(\frac{1}{\mathcal{X}}\right)$$

Setting initial conditions:

$$F_0 = 0$$

$$F_1 = 1$$

$$F_2 = \Pi$$

It follows, after simplification:

$$\alpha = -\beta$$

$$\alpha = \frac{2(\pi - 3)(7 - 2\pi)}{(8\pi^2 - 52\pi + 85)} \approx 0.341726437$$

$F_k$  can consequently be expressed, according to k index and  $\mathcal{X}$  ratio as follows:

$$\Rightarrow F_k = U(k) = \frac{2(\pi - 3)(7 - 2\pi)}{(8\pi^2 - 52\pi + 85)} \left[ \mathcal{X}^k - \left(-\frac{1}{\mathcal{X}}\right)^k \right] \quad \text{where } \mathcal{X} = \frac{7 - 2\pi}{2(\pi - 3)}$$

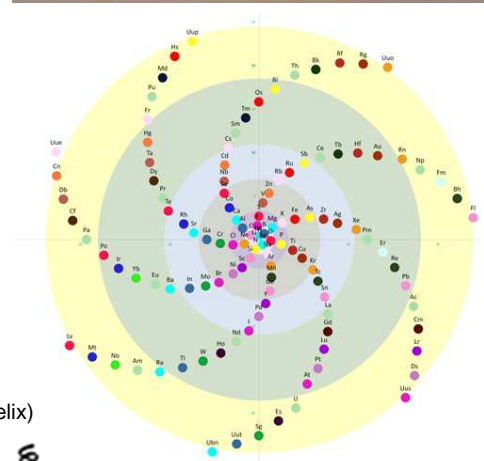
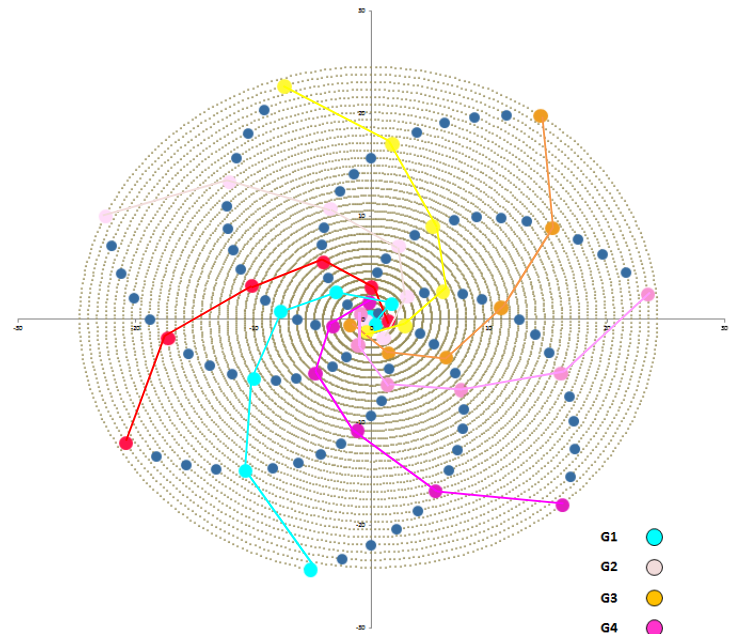
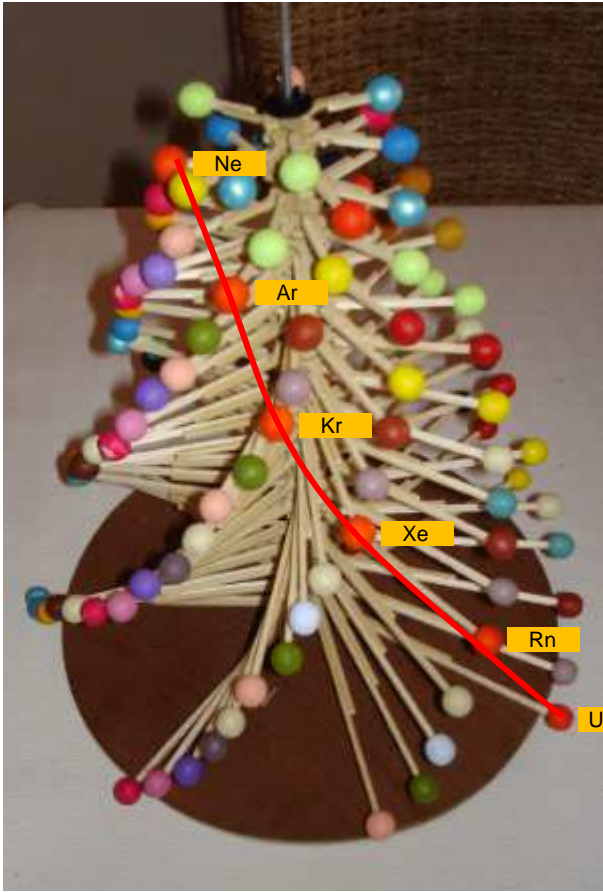


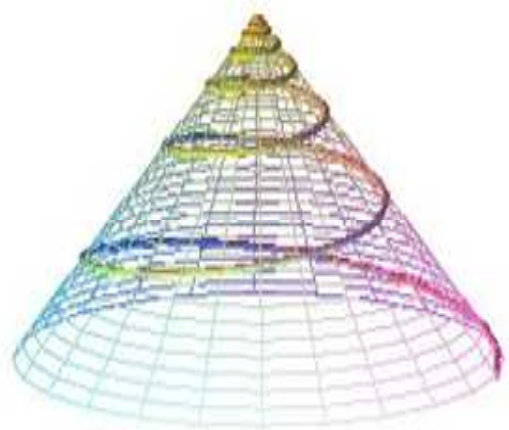
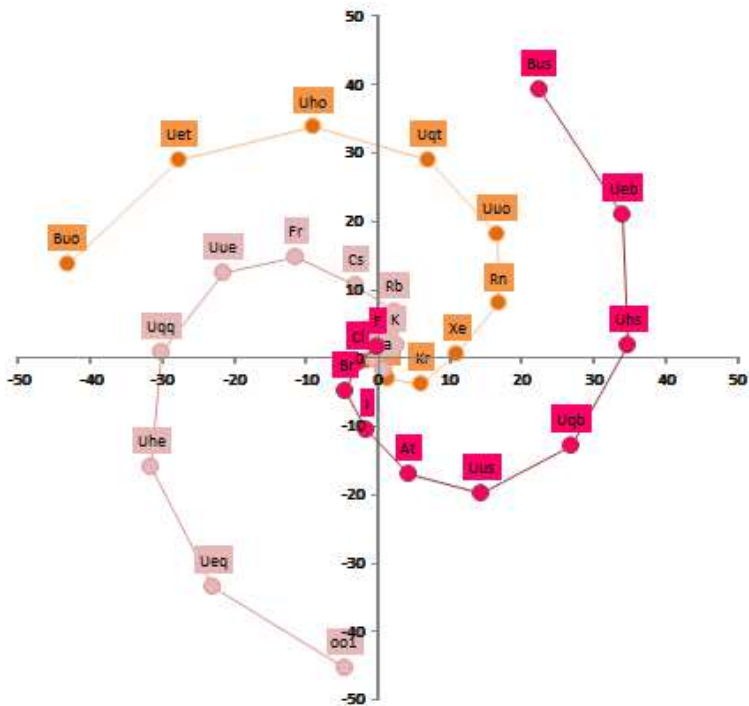
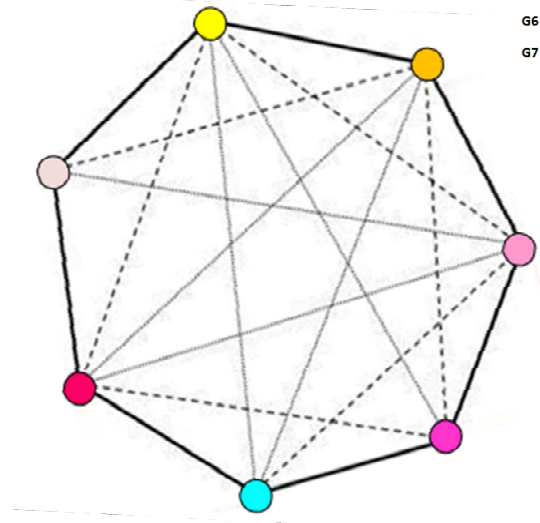
Figure 8

Initial and final arrangement mock-up (conical helix)

$$\gamma_{Deg} \approx 101.9467106^\circ \approx 102^\circ = \omega$$



- G1 ●
- G2 ●
- G3 ●
- G4 ●
- G5 ●
- G6 ●
- G7 ●



## Ж and Powers of Ж:

index	Sequence	Ж	logЖ
1	1	1	0
2	Ж	2.531256653	0.403336
3	$Ж^2 = \pi * Ж + 1$	6.407260243	0.806672
4	$Ж^3 = \pi * Ж^2 + Ж = \pi * \pi * Ж + \pi + Ж = (\pi^2 + 1) * Ж + \pi$	16.21842012	1.210009
5	$Ж^4 = (\pi^2 + 1) * Ж^2 + \pi * Ж = \pi(\pi^2 + 2) * Ж + \pi^2 + 1$	41.05298382	1.613345
6	Ж^5=...	103.9156384	2.016681
7		263.0371511	2.420017
8		665.8145387	2.823353
9		1685.347481	3.226689
10		4266.047023	3.630026
11		10798.45991	4.033362
12		27333.67349	4.436698
13		69188.54287	4.840034
14		175133.9594	5.24337
15		443309	5.646707
16		1122128.856	6.050043
17		2840396.131	6.453379
18		7189771.604	6.856715
19		18199157.21	7.260051
20		46066737.76	7.663387
21		116606736.4	8.066724
22		295161577.4	8.47006

## Spiral arrangement in numbers:

The below figure shows the distribution of elements in spiral shape with the successive layers, elements names, and their atomic numbers.

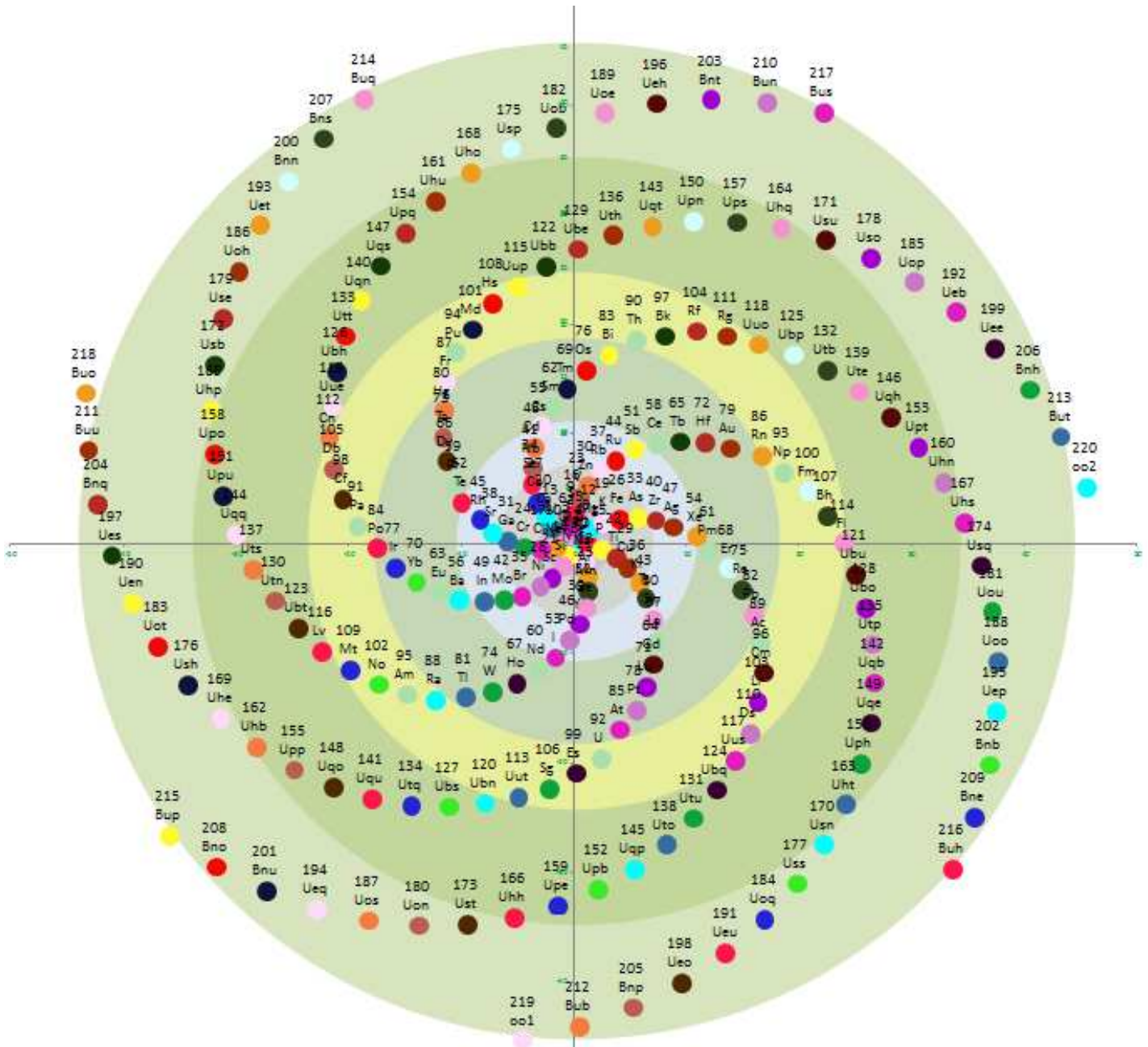


Figure 9

Spiral arrangement of elements with their Atomic numbers and names

# ANNEXE Standard form of the periodic table

(two standard representations as grids of elements with the periods in rows and groups in columns, and used as the baseline for the calculations and definition of the above 7-arms spiral arrangement of elements).

### Tableau périodique des éléments

**Annotations for Iron (Fe):**

- Nom de l'élément: Fer
- Numéro atomique: 26
- Symbole de l'élément (en gras : aucun isotope stable): Fe
- Masses atomiques, basée sur  $^{12}\text{C}$ : 55.845
- Énergie de ionisation (eV): 7.62
- Énergie de premier ionisation (eV): 7.62
- Configuration électronique (en gras : accepteur à la règle de Hund):  $[\text{Ar}] 3d^6 4s^2$
- Plusieurs autres propriétés (à compléter au plus haut niveau possible)

**Legend:**

- metaux alcalins
- metaux alcalino-terreux
- autres métaux
- métaux de transition
- metalloïdes
- non-métaux
- halogènes
- gaz nobles
- éléments inconnus
- les éléments indiqués en jaune sont des éléments synthétiques

\* Pure Appl. Chem., Vol. 78, No. 11, pp. 2051-2066, 2006. Actualisé en 2013 selon recommandations de l'Union internationale de Chimie Pure et Appliquée. © 2013, Cova Dargan - Astra-Science / www.dargan.net - www.astra-science.ro

### Tableau périodique des éléments chimiques

**Annotations for Iron (Fe):**

- numéro atomique: 26
- nom: Fer
- configuration électronique:  $[\text{Ar}] 3d^6 4s^2$
- catégorie: métal de transition

**notes**

- pour l'atome, les éléments 113, 115, 117 et 118 n'ont pas de nom officiel désigné par l'IUPAC
- 1 kg/mol = 96,485 kJ
- tous les éléments sont impliqués dans des états d'oxydation de 0.

**boîtes de configuration électronique**